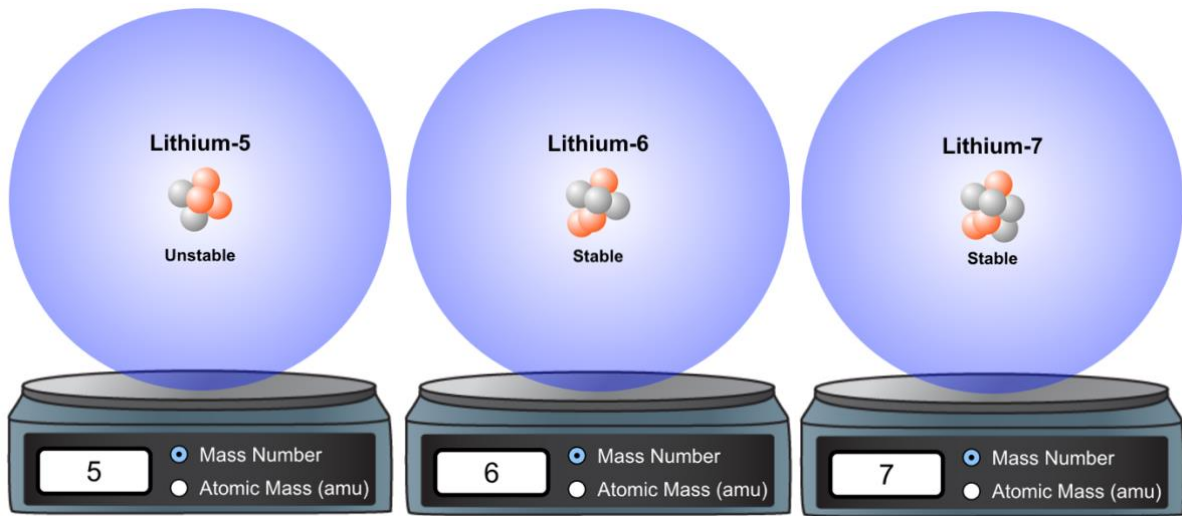
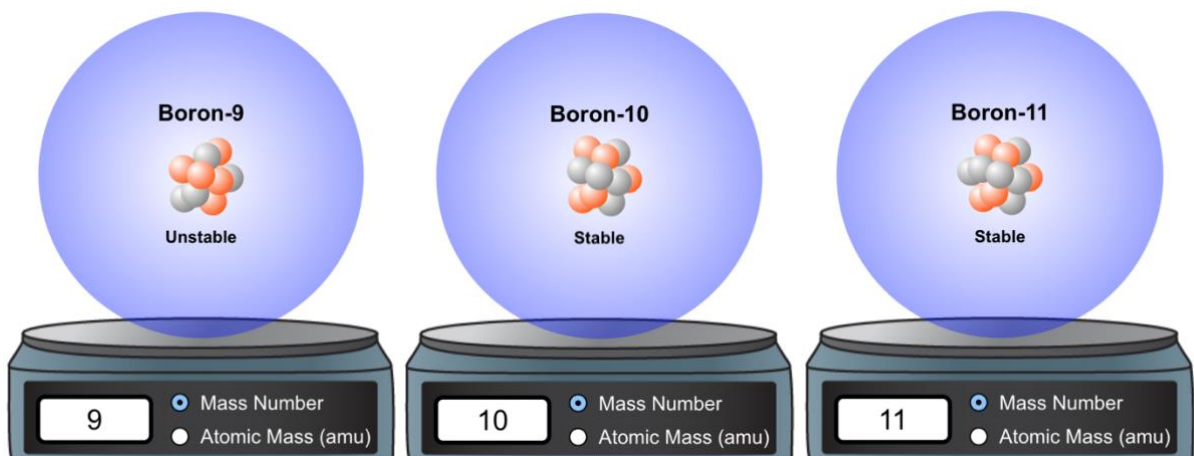


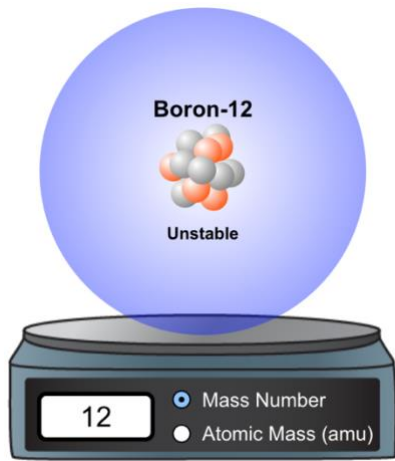
Isotope



Lithium:

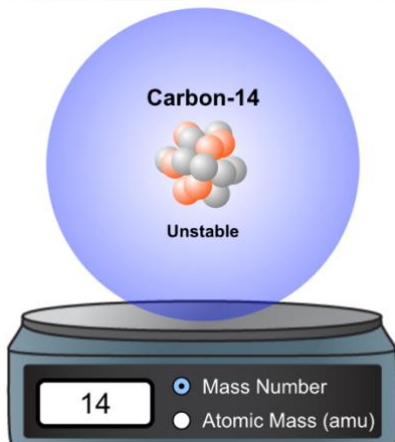
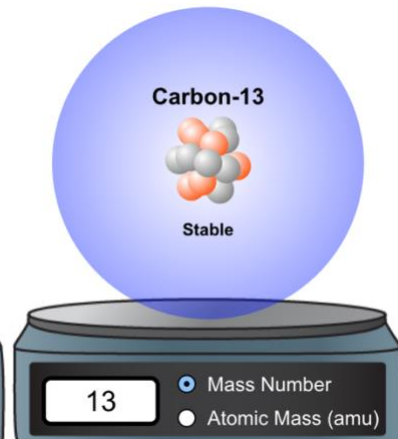
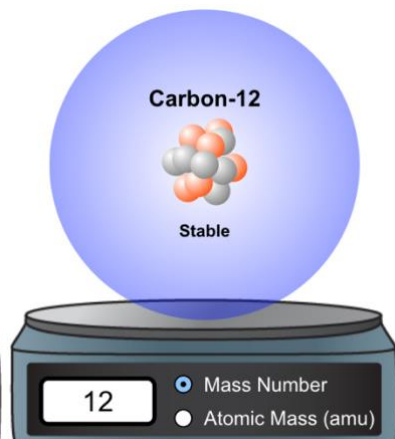
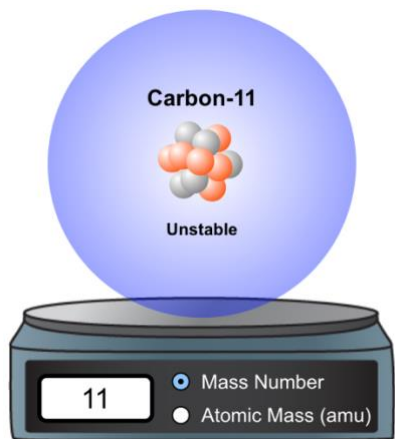
- Immer 3 Protonen
- Nur Lithium-6 u. Lithium-7 stabil
- Zahl hinter Lithium- zeigt die Masse
- Masse – Protonen = Neutronen
- Immer 3 Elektronen





Boron:

- Immer 5 Protonen
- Nur Boron-10 u. Boron-11 stabil
- Zahl hinter Boron- zeigt die Masse
- Masse – Protonen = Neutronen
- Immer 5 Elektronen



Carbon:

- Immer 6 Protonen
- Nur Carbon-12 u. Carbon-13 stabil
- Zahl hinter Carbon- zeigt die Masse
- Masse – Protonen = Neutronen
- Immer 6 Elektronen